

Thermodynamics of Imidazolium-Based Ionic Liquids Containing the Trifluoromethanesulfonate Anion

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Abstract

© 2018 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim A thermochemical investigation of the series of 1-alkyl-3-methylimidazolium ionic liquids (ILs) with the trifluoromethanesulfonate CF₃SO₃[−] anion is presented. Absolute vapor pressures and vaporization enthalpies were measured by using a quartz crystal microbalance (QCM). The enthalpies of solutions were determined by solution calorimetry. Gas-phase enthalpies of the formation of ILs were calculated by applying the high-level quantum-chemical method G3MP2. From a combination of experimental and theoretical results, the enthalpy of formation of aqueous CF₃SO₃[−] anion was derived for the first time.

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Keywords

Enthalpy of formation, Enthalpy of vaporization, Ionic liquids, Thermodynamics, Vapor pressure

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